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10/045,391	11/09/2001	Sheng-Shing Li	PP/1-22278/P5/CGC 2069	2361
Patent Departm	7590 01/10/200 <sup>o</sup>	EXAMINER		
Ciba Specialty Chemicals Corporation 540 White Plains Road P.O. Box 2005 Tarrytown, NY 10591-9005			CHOI, PETER Y	
			ART UNIT	PAPER NUMBER
			1771	
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U.S. Patent and Trademark Office PTOL-326 (Rev. 08-06)

Paper No(s)/Mail Date

2) Notice of Draftsperson's Patent Drawing Review (PTO-948)

3) Information Disclosure Statement(s) (PTO/SB/08)

Paper No(s)/Mail Date. \_\_\_\_

6) Other: \_

5) Notice of Informal Patent Application

#### NON-FINAL ACTION

1. Applicants' remarks of October 26, 2006, have been received and are persuasive as to the 35 U.S.C. 102(e) and 35 U.S.C. 103(a) rejections based on USPN 6,146,757 to Mor.

### Claim Rejections - 35 USC § 103

- 2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 3. Claims 1, 2, 7-12, 17-19, and 25 are rejected under 35 U.S.C. 103(a) as being unpatentable over USPN 5,348,736 to Patel in view of USPN 6,146,757 to Mor.

Claims 1, 2, 10, 11, and 17-19 remain rejected as substantially set forth in the Non-Final Rejection of July 24, 2006, section 5.

Regarding claims 7 and 8, Patel does not appear to teach a polyolefin fiber or filament wherein the compounds of component (b) are present from about 0.1% to about 15%, or from about 1% to about 7%, by weight based on the polyolefin of component (a). Since Patel is silent with regards to specific materials, it would have been necessary and thus obvious to look to the prior art for conventional materials. Mor provides this conventional teaching showing that it is known in the art to combine a wetting agent with polyolefin fibers, the fibers or filaments containing about 1 to about 20 percent by weight of a combination of the first wetting agent and a second wetting agent (column 7 lines 53-67, column 8 lines 1-5, column 9 lines 65-67). Therefore, it would have been obvious to one having ordinary skill in the art at the time the

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invention was made to form a polyolefin fiber or filament of Patel comprising from about 0.1% to about 15%, or from about 1% to about 7%, of component (b), as taught by Mor, motivated by the expectation of successfully practicing the invention of Patel.

Regarding claims 9 and 12, Patel does not appear to teach a bicomponent fiber comprising a polyolefin component wherein the polyolefin component comprises a melt blend comprising components (a) and (b), and a woven or nonwoven fabric comprising bicomponent fibers. Since Patel is silent with regards to the specific composition of the fabric, it would have been necessary and thus obvious to look to the prior art for conventional materials. Mor provides this conventional teaching showing that it is known in the art to combine a wetting agent with polyolefin fibers, where the composition of the fibers and wetting agent can be a blend component for other fibers in a woven, nonwoven or knitted fabric, made of preferably a polyolefin such as polyethylene or polypropylene (column 13 lines 28-33). Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to make the fibrous material in Patel with bicomponent fibers to form a nonwoven or woven fabric, as taught by Mor, motivated by the expectation of successfully practicing the invention of Patel.

Regarding claim 25, Patel does not appear to teach a fiber or filament further comprising a stabilizer selected from the group consisting of hindered amine light stabilizers, phenolic antioxidants, phosphites or phosphonites, hydroxylamines, benzofuranones and hydroxyphenylbenzotriazole, hydroxybenzophenone or tris-aryls-s-triazine UV absorbers. However, Mor teaches adding a phosphite antioxidant and stabilizer to permit high temperature processing while not materially interfering with the properties of the wetting agent in the composition (column 10 lines 25-53). Therefore, it would have been obvious to one having

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ordinary skill in the art at the time the invention was made to add the phosphite antioxidant motivated by the desire to make a fabric which will be more durable when processed without materially interfering with the properties of the wetting agent.

### Response to Arguments

4. Applicants' arguments filed October 26, 2006, have been fully considered but they are not persuasive. Applicants have submitted remarks and a Rule 132 Declaration by Dr. Gande comparing UNITHOX 420 to UNITHOX 480 and 750, arguing that UNITHOX 420, which is equivalent to the claimed compound (b), displays superior water absorption or wettability.

Examiner respectfully disagrees. First, Patel teaches using an ethoxylated alcohol, which will normally contain between 18 to 54 carbon atoms and up to about 20 ethoxy groups per mole (columns 3 and 4). The preferred alcohol is UNILIN 425 which comprises about 30 carbon atoms (column 3 lines 64-67). Additionally, the derivatives of the UNILIN's may be employed (column 4 lines 11-20). While Patel does not teach the exact structure of the ethoxylated UNILIN 425, the non-patent literature entitled "The Use of UNILIN Alcohols in the Formation of Ethoxylates and Their Properties" by the Petrolite Specialty Polymers Group, teaches that a derivative of UNILIN 425 ethoxylate may contain 10-80% of ethylene oxide to form the ethoxylate alcohol consisting of 1, 2, or 4 monomers of ethoxy groups. As Patel teaches that the ethoxy groups may comprise up to about 20 ethoxy groups per mole, and the non-patent literature teaches that a derivative of UNILIN 425 may contain 1, 2 or 4 monomers of ethoxy groups, UNILIN 425 is deemed to read on the claimed invention.

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Second, the Rule 132 Declaration by Dr. Gande is not persuasive. Applicants argue that UNITHOX 420 has superior liquid absorption capacity when compared to UNITHOX 480 and UNITHOX 750. However, neither of these ethoxylated alcohols are taught nor disclosed in the prior art. As ethoxylated UNITHOX 425 contains about 30 carbon atoms and may contain 1, 2, or 4 ethoxy groups, or up to about 20 ethoxy groups as taught in the prior art, neither UNITHOX 480 nor UNITHOX 750 is analogous to the exthoxylated UNITHOX 425. UNITHOX 480 exceeds the number of ethoxy groups and UNITHOX 750 exceeds the number of carbon atoms. As set forth above, the ethoxylated derivative of UNITHOX 425 appears to be analogous to the claimed UNITHOX 420. Therefore, the Declaration is not persuasive as it does not show that UNITHOX 420 has superior liquid absorption capacity when compared to UNITHOX 425.

#### Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Peter Y. Choi whose telephone number is (571) 272-6730. The examiner can normally be reached on Monday - Friday, 08:00 - 15:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Terrel Morris can be reached on (571) 272-1478. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Peter Y. Choi January 3, 2007

ANDREW PIZIALI
PRIMARY EXAMINER